CHJ-7201

Seat No.\_\_\_\_

M. Sc. (Sem. I) Examination November - 2019

Chemistry: Paper - IV - CHNN - 401

("Inorganic Chem.") (New Course)

Time :  $2\frac{1}{2}$  Hours

(Total Marks: 70

**Instruction**: All questions are compulsory.

- Answer any two of the following in detail: 18
  - Explain the V.S.E.P.R. theory with appropriate examples.
  - Explain the atomic inversion by proper illustration.
  - Write short notes on:
    - $d\pi p\pi$  bonding
    - Bent rule.
- Answer in detail: (any two)

17

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- (1) What is chelate effect? Explain the chelate effect with appropriate example and thermodynamic origin for metal complexes.
- Explain the spectrophotometric method for determination of binary formation constant.
- Explain the factors affecting the stability of the metal complexes.

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- Answer any two of the following in brief:
  - (1) Explain the innert and labile complexes with taking appropriate examples.
  - (2) Explain the electron transfer reactions.
  - What is trans effect? Explain the substitution reaction in square planner complexes of Pt.
- Answer any two of the following:
  - Explain the M.O. theory for octahedral complexes of CO. (III).
  - Explain the Tanabe Sugano diagram for the  $d^2$  case.
  - Calculate the C.F.S.E. value for the following complexes:

(a) 
$$\left[Fe(CN)_{6}\right]^{-3} (z = 26)$$
  
(b)  $\left[CoF_{6}\right]^{-3} (z = 27)$   
(c)  $K_{4}\left[Fe(CN)_{6}\right]$ 

(b) 
$$[CoF_6]^{-3}$$
  $(z=27)$ 

(c) 
$$K_4\left[Fe(CN)_6\right]$$

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